Alkaloids From Roots of Alseodaphne Corneri Kosterm

¹Mohd Azlan Nafiah, ²Mat Ropi Mukhtar, ³Hiroshi Morita, ¹Kartini Ahmad, ²Khalijah Awang, ²A. Hamid A. Hadi

¹Department of Chemistry, Faculty of Science and Technology, University Pendidikan Sultan Idris, 35900, Tg. Malim, Perak.

²Department of Chemistry, Faculty of Science, University Malaya, 50603 Kuala Lumpur. ³Faculty of Pharmaceutical Sciences, Hoshi University, Shinagawa-ku, Tokyo 142-8501, Japan Received in 15th September 2010, accepted in revised form 25th November 2010.

ABSTRACT This is the first report on the occurrence of bisbenzylisoquinoline alkaloids in *Alseodaphne corneri* which belongs to the family of Lauraceae. Chemical studies on the roots of this species have yielded four bisbenzylisoquinoline alkaloids; (-)-gyrolidine 1, norstephasubine 2, (+)-2-norlimacusine 3 and (+)-stephasubine 4. The isolation was achieved by chromatographic techniques and the structural elucidation was performed via spectral methods; namely 1D and 2D NMR, IR, UV and MS, and in comparison with published literature.

(Keywords: bisbenzylisoquinoline, *alseodaphne corneri*, spectroscopy)

INTRODUCTION

Alseodaphne corneri Kosterm (KL 4928) of Lauraceae, grows as a wild plant, 6-8 m high. The plant is also known as Medang and the genus includes more than 50 species, distributed through the Yunnan to West Malaysia and 23 species are found in Malaysia [1]. This paper reports the isolation and identification of four bisbenzylisoquinoline from root extract of the plant species. Structural elucidation was performed with the aid of spectroscopic methods; ¹H/¹³C-NMR, IR, UV, MS.

MATHERIAL AND METHODS

Plant material

۶f

The roots of *Alseodaphne corneri* were obtained from University Malaya Herbarium. 3 Kg of the air-dried roots of *Alseodaphne corneri* were moistened with 25% NH₄OH and soaked in CH₂Cl₂ for 3 days (cold extraction). The CH₂Cl₂ extract was evaporated to 500ml followed by extraction using 5% HCl until Mayer's test is negative. The HCl extract was basified with concentrated ammonia to pH 11 and reextracted

with CH_2Cl_2 . The CH_2Cl_2 was washed with distilled H_2O and dried over anhydrous sodium sulphate. Finally, the extract was evaporated to dryness to give crude alkaloid (6.8g). The extract was subjected to the gradient elution column chromatography, using mixtures of hexane, hexane/ CH_2Cl_2 , CH_2Cl_2 and CH₂Cl₂/MeOH as eluants. A total of 90 fractions were obtained and fraction 29-30 were further purified using the preparative TLC (Silica gel 60 F_{254} , CH₂Cl₂:MeOH; 98:2, 97:3, 96:4) afforded (-)-gyrolidine 1, norstephasubine 2, (+)-2-norlimacusine 3 and (+)-stephasubine 4, respectively.

(-)-Gyrolidine (1): UV λ_{max} (MeOH) nm: 261 and 282; IR υ_{max} cm $^1\!\!:$ 3401 (OH), 1507 and 1228. Mass spectrum m/e (%): 622; $[\alpha]_{\rm p}^{26}$ -115° (c 1.1, MeOH); ¹H NMR (CDCl₃) ppm: 3.88 (3H, s, 12-OCH₃), 3.78 (3H, s, 6'-OCH₃), 3.62 (3H, s, 6-OCH₃), 3.18 (3H, s, 7'-OCH₃), 2.66 (3H, s, N'-CH₃), 2.57 (3H, s, N-CH₃), 7.42 (1H, dd, J= 8.2 and 1.4 Hz, H-14'), 6.95 (1H, d, J= 6.95 Hz, H-13'), 6.93 (1H, d, J= 2.3 Hz, H-10'), 6.80 (1H, d, J= 9.0 Hz, H-14), 6.76 (1H, d, J= 8.2 Hz, H-13), 6.63 (1H, s, H-8), 6.37 (1H, d, J= 5.4 Hz, H-11'), 6.35 (1H, s, H-5'), 6.31 (1H, s, H-5), 5.44 (1H, d, J= 1.8 Hz, H-10), 4.21 (1H, d, J= 5.4 Hz, H-1'), 3.64 (1H, m, H-1), 2.42 (1H, m, H_{ax}-3), 2.76 (1H, m, Heq-3), 2.34 (2H, m, Hax-4 and Heq-4), 2.86 (1H, dd, J= 14.6 and 3.6 Hz, $H_{ax}-\alpha$), 3.15 (1H, m, $H_{ea}-\alpha$), 4.21 (1H, d, J= 5.4 Hz, H-1'), 2.93 (1H, m, H_{ax}-3'), 3.21 (1H, m, H_{eq} -3'), 2.70 (1H, m, H_{ax} -4'), 3.03 (1H, m, $H_{ea}-4'$), 2.80 (1H, dd, J= 14.6 and 5.9 Hz, $H_{ax}-\alpha'$), 3.35 (1H, d, J= 17.8 Hz, $H_{eq}-\alpha'$).

Norstephasubine (2): UV λ_{max} (MeOH) nm: 240, 286 and 338; IR υ_{max} cm⁻¹: 2950 (*N*-H). Mass spectrum m/e (%): 576; $[\alpha]_{D}^{26}$ +309° (*c* 1.0, MeOH); ¹H NMR (CDCl₃) ppm: 3.87 (3H, *s*, 12-OCH₃), 4.04 (3H, *s*, 6'-OCH₃), 4.05 (3H, *s*, 6-

·281



OCH₃), 8.42 (1H, d, J= 8.4 Hz, H-3'), 7.46 (1H, d, J= 5.6 Hz, H-4'), 7.35 (1H, d, J= 7.0 Hz, H-14'), 7.04 (1H, d, J= 8.0 Hz, H-10'), 6.97 (1H, s, H-5'), 6.71 (2H, s, H-14 and H-13), 6.64 (1H, dd, J= 1.9 and 8.2 Hz, H-11'), 6.55 (1H, s, H-5), 6.43 (1H, dd, J= 1.9 and 8.2 Hz, H-13'), 6.04 (1H, s, H-8), 4.89 (1H, d, J= 1.7 Hz, H-10), 5.37 (1H, d, J= 13.9 Hz, H-a'), 4.51 (1H, d, J= 13.6, H-a'), 4.08 (1H, d, J= 3.9 Hz, H-1), 2.55 (1H, m, H_{ax}-3), 2.92 (1H, d, J= 15.8 Hz, H_{eq}-3), 2.70 (1H, m, H_{ax}-a), 2.73 (1H, m, H_{eq}-a).

(+)-2-Norobaberine (3): UV λ_{max} (MeOH) nm: 240, 286 and 338; IR υ_{max} cm⁻¹: 3401. Mass spectrum m/e (%): 576; [α] $_{\rm D}^{26}$ -170° (*c* 1.0, MeOH); ¹H NMR (CDCl₃) ppm: 7.46 (1H, *d*, *J*= 8.0 Hz, H-14'), 6.96 (1H, *d*, *J*= 8.0 Hz, H-13'), 6.95 (1H, *d*, *J*= 8.0 Hz, H-11'), 6.90 (1H, *d*, *J*= 8.0 Hz, H-10'), 6.84 (1H, *d*, *J*= 8.0 Hz, H-13), 6.79 (1H, *d*, *J*= 8.0 Hz, H-14), 6.67 (1H, *s*, H-8), 6.34 (1H, *s*, H-5'), 6.33 (1H, *s*, H-5), 5.57 (1H, *br s*, H-10), 4.24 (1H, *br s*, H-1'), 4.21 (1H, *br s*, H-1), 3.89 (3H, *s*, 12-OCH₃), 3.76 (3H, *s*, 6'-OCH₃), 3.60 (3H, *s*, 6-OCH₃), 3.18 (3H, *s*, 7'-OCH₃), 2.66 (3H, *s*, *N'*-CH₃).

(+)-Stephasubine (4): UV λ_{max} (MeOH) nm: 240, 287 and 337; IR υ_{max} cm⁻¹: 3400, 1460. Mass spectrum m/e (%): 590; $[\alpha]_D^{26}$ +339° (*c* 0.09, MeOH); ¹H NMR (CDCl₃) ppm: 4.04 (3H, *s*, 6'-OCH₃), 4.01 (3H, *s*, 6-OCH₃), 3.83 (3H, *s*, 12-OCH₃), 2.47 (3H, *s*, *N*-CH₃), 8.40 (1H, *d*, *J*= 5.6 Hz, H-3'), 7.44 (1H, *d*, *J*= 5.3 Hz, H-4'), 7.38 (1H, *d*, *J*= 7.0 Hz, H-14'), 6.95 (1H, *s*, H-5'), 6.94 (1H, *d*, *J*= 8.5 Hz, H-10'), 6.67 (1H, *s*, H-13), 6.60 (1H, *d*, *J*= 8.0 Hz, H-13'), 5.95 (1H, *s*, H-8),



4.74 (1H, s, H-10), 5.33 (1H, d, J= 13.9 Hz, $H_{eq}-\alpha'$), 4.47 (1H, d, J= 14.1 Hz, $H_{ax}-\alpha'$), 3.64 (1H, br s, H-1), 2.44-2.53 (2H, m, H-3), 2.98 (1H, dd, J= 14.1 and 3.6 Hz, $H_{eq}-\alpha$), 2.23 (1H, m, $H_{ax}-\alpha$), 2.42 (1H, m, $H_{eq}-4$), 2.18 (1H, m, Hax-4).

RESULTS AND DISCUSSION

Compound (1) was isolated as yellow amorphous from CH₂Cl₂. the UV spectrum showed bands at λ_{max} (MeOH) 261 and 282 nm which showed typical of bisbenzylisoquinoline moiety [2, 3]. The IR spectrum revealed the presence of OH group. Peaks at 1507 cm⁻¹ indicated the presence of conjugated doubled bond, while peaks at 1228 cm⁻¹ corresponded to C-O group [4]. Mass spectrum of the compound gave molecular at m/e 622 that corresponded to molecular formular C₃₈H₄₂N₂O₆.

The ¹H NMR spectrum displayed mutually coupled signals at δ 7.44 (J= 5.4 Hz) and δ 8.40 (J= 5.6 Hz) due to the presence of a substituted pyridine system. Conspicuously present were two doublets at δ 4.47 and δ 5.33, with a large coupling constant at 14.1 Hz and 13.9 Hz, respectively, which represented the two geminal protons of the benzylic methylene adjacent to the pyridine ring. The presence of H-1 broad singlet upfield at δ 3.64, accompanied by an Nmethyl signal at 82.47, argued convincingly in favor of placing the pyridine system on the right hand-side of the dimmer [5, 6]. The peaks for three methoxyls singlet appeared at δ 4.00, 4.01 and 3.83 corresponding to C-6', C-6 and C-12 respectively. The resonances for H-10', H-11', H-13' and H-14' appeared as a doublet at δ 6.94 (J = 8.0 Hz), 6.60 (J =

8.5 Hz), 6.45 (J = 8.0 Hz) and 7.38 (J = 8.0 Hz). Six singlet aromatic protons (each 1H) observed at δ 4.74, 5.95, 6.51, 6.67, 6.67 and 6.94 were attributable to H-10, H-8, H-5, H-13, H-14 and H-5', respectively. Based on the above data, compound (1) was elucidated as (-)-gyrolidine [7] which were from VI type.

The second alkaloid, compound (2) was isolated as pale yellow amorphous solid. Its UV spectrum showed absorption bands at λ_{max} (MeOH) 240, 286 and 338 nm typical of the conjugated quinonoid moiety [8]. In addition, the IR spectrum gave broad band at 2950 cm⁻¹ due to presence of C-H group.

The ¹H NMR spectrum exhibited a relative symmetry of the chemical shift pattern of the aromatic protons resonated as doublet signals at δ 8.42 and 7.46 corresponding to H-3' and H-4', respectively, with coupling constant J = 5.6 Hz which is an *ortho* disubstituted aromatic ring. The former peak was assigned in the downfield region due to the fact that it was attached to the carbon adjacent to nitrogen atom and as a result it was more deshielded compared to the latter. Another set of doublet signals presence at δ 7.35 (J= 7.0 Hz) and 7.04 (J= 8.0 Hz) corresponding to two aromatic protons, H-14' and H-10'. H-13' and H-11' were resonated as two doublet doublets signals at δ 6.43 (J= 2.4 and 8.2 Hz) and 6.64 (J= 1.9 and 8.2 Hz), respectively. A doublet which consists of two protons resonated at δ 6.72 with coupling constant 8.3 Hz referred to H-14 and H-13. In addition, a doublet signal with small coupling constant 1.7 Hz at high field region δ 4.89 were referred to H-10. The ¹H- NMR also showed three singlet signals at δ 6.97, 6.55 and 6.04 corresponding to H-5', H-5 and H-8, respectively. In addition, another three singlet signals at δ 4.05, 4.04 and 3.87 were referred to three methoxyl groups at C-6, C-6' and C-12, respectively. On the basis of the spectroscopic studies and data obtained from the literature reviews alkaloid (2) (VI type) was assigned as a norstephasubine [9].

ć

S

f

n

7

d

)

e

ır -

d

z)

۱.

7

z

0

lt

ιd

√-

or

le

ls

33

y.

4'

=

The third alkaloid, compound (3) was isolated as yellow amorphous from CH₂Cl₂. The UV spectrum showed absorption bands at λ_{max} (MeOH) 295 nm. The IR spectrum showed absorption at 3401cm⁻¹ indicating the presence of *N*-H group. The mass spectrum exhibited a molecular ion peak at m/e 608 suggesting a molecular formula of C₃₇H₄₀N₂O₆. Other significant fragmentation peaks were revealed at m/e 577 [M-31]⁺ indicating the loss of methoxyl group.

The ¹H-NMR spectrum showed four methoxyl singlet signals at δ 3.18, 3.60, 3.76 and 3.89 attached to C-

7', C-6, C-6' and C-12, respectively. The singlet proton signal at δ 2.66 indicated presence of the methyl group at B' ring attached to nitrogen atom. Six doublet aromatic protons were observed at δ 7.46, 6.96, 6.95, 6.90, 6.84 and 6.79 with coupling constant 8.0 Hz which attributable to H-14', H-13', H-11', H-10', H-13 and H-14, respectively. Another 4 aromatic protons were observed as a singlet at δ 6.67 (H-8), 6.34 (H-5'), 6.33 (H-5) and 5.57 (H-10). In addition two broad singlet signals which attributable to H-1 and H-1' were observed at δ 4.21 and 4.24, respectively. Comparison of the empirical data with the literature values of the known compound [8, 10] was deduced as (+)-2-norobaberine (VI type).

The last alkaloid, compound (4) was isolated as yellow amorphous. Its tend to darken when exposed to air or light which showed unstable condition of the compound. Alkaloid (4) showed a peak at 3400 cm⁻¹ typical of the stretching of a hydroxyl group [4]. The other significant peaks were also observed at 1460 cm⁻¹ which showed the presence of the imine chromophore of a dihydroisoquinoline moiety [5]. The UV spectrum display maxima absorption at 337 nm confirmed the presence of bisbenzylisoquinoline moiety [2]. The 'H-NMR spectrum displayed mutually coupled signals at δ 7.44 (J= 5.4 Hz) and δ 8.40 (J= 5.6 Hz) due to the presence of a substituted pyridine system. Conspicuously present were two doublets at δ 4.47 and δ 5.33, with a large coupling constant at 14.1 Hz and 13.9 Hz, respectively, which represented the two geminal protons of the benzylic methylene adjacent to the pyridine ring. The presence of H-1 broad singlet upfield at δ 3.64, accompanied by an N-methyl signal at $\delta 2.47$, argued convincingly in favor of placing the pyridine system on the right hand-side of the dimmer [5, 6]. The peaks for three methoxyls singlet appeared at δ 4.00, 4.01 and 3.83 corresponding to C-6', C-6 and C-12 respectively. The resonances for H-10', H-11', H-13' and H-14' appeared as a doublet at δ 6.94 (J= 8.0 Hz), 6.60 (J = 8.5 Hz), 6.45 (J = 8.0 Hz) and 7.38 (J = 8.0 Hz). Six singlet aromatic protons (each 1H) observed at \delta 4.74, 5.95, 6.51, 6.67, 6.67 and 6.94 were attributable to H-10, H-8, H-5, H-13, H-14 and H-5', respectively. As a result, the author deduced that compound (4) is indeed (+)-stephasubine which isolated from species of Stephania suberosa Forman [11].

ACKNOWLEDGEMENT

The study was supported by University Malaya Research Grant (PS147/2007B). The author also wishes to thank to Teo Leong Eng, Pak Din, Hazry and Rafly for their excellent technical assistance.

REFERENCES

- Ng, F.S.P., *Tree Flora of Malaya, A Manual for Foresters*. Vol. 4. 1989: Longman Malaysia Sdn. Bhd. 109-117.
- Sangster, A.W. and K.L. Stuart, (1965). Chemical Reviews, 65: p. 69-130.
- Kanyinda, B., R. Vanhaelen-Fastre, M. Vanhaelen and R. Ottinger, (1997). Two New Isochondodendrine-Type Alkaloids from the Roots of *Anisocycla jollyana.*, *Journal of Natural Products*, 60: p. 1121-1124.
- Williams, D.H. and I. Fleming, Spectroscopic Methods in Organic Chemistry. 4th Ed ed. 1989: McGraw-Hill Book Company.
- Amarendra, P., K.M. Tarun, K.M. Prabir and C.R. Brindaban, (1988). (+)-3',4'-dihydrostephasubine, a bisbenzylisoquinoline alkaloid from *Stephania hernandifolia*, *Phytochemistry*, 27(2): p. 653-655.
- 6. Guinaudeau, H., M. Lebeouf and A. Cave, (1975). *Lloydia*, **38**: p. 294.

- Chalandre, M.C., J. Bruneton, P. Cabalion and H. Guinaudeau, (1986). Alcaloides de Gyrocarpus americanus Journal of Natural Products, 49(1): p. 101-105.
- Damas, P., J. Bruneton, A. Fournet and H. Guinaudeau, (1985). (+)-3',4'dihydrostephasubine, a bisbenzylisoquinoline alkaloid from *Stephania hernandifolia*, *Journal of Natural Products*, 48(1): p. 69-71.
- Patra, A., Freyer, A. J., Guinaudeau, H, Shamma, M., Tantisewie, B., Pharadai, K., (1986). The Bisbenzylisoquinoline Alkaloids is *Stephania* suberosa., Journal of Natural Products 49: p. 424.
- Bamrung, T., A. Susan, G. Helene and S. Maurice, (1989). New Bisbenzylisoquinoline from Stephania pierrii, Journal of Natural Products, 52(4): p. 846-851.
- Patra, A., (1987). Promorphinane and hasubanane alkaloids of *stephania suberosa*, *Phytochemistry* 26: p. 2391.

5

l

ERRATUM [Vol 29 (1)]

Inadvertently the following figures and tables were missed in the final printing for paper by Kumar D. et al. (page 52 to 61).











and feed ingredients
if bio wastes*
composition of
Bio chemical
Table 1:

Ingredient	Protein (%)	Carbohydrate %	Lipid (%)	Ash (%)	GE/kcal/100g ^a	E/P**
Chicken intestine*	68.45	3.93	• 10.12	15.72	407.70	5.95
Fish waste*	58.06	3.69	6.18	24.21	326.59	5.62
Silkworm pupae*	55.02	1.00	17.12	22.01	396.73	7.21
Soya bean meal	49.64	9.20	9.70	5.20	337.66	6.80
Ground nut oilcake	48.04	6.90	10.9	11.64	332.74	6.92
Rice bran	15.90	20.4	3.90	7.90	175.11	11.01
Tapioca	14.60	43.7	0.20	8.60	218.16	14.94

Note:

*bio waste

**Energy/protein

^a Gross energy

. . . .

Table 2: Percentage and proximate composition of formulated diets

		*	Crude protein level %	9	
Ingredients	40	45	50	55	09
Chicken intestine	13.8	22.9	32.8	40.0	47.0
Fish waste	18.0	18.0	16.8	20.0	20.0
Silk worm pupae	19.0	19.0	20.0	20.0	20.0
Ground nut oil cake	6.0	6.0	6.0	6.0	3.0
Soya bean flour	5.0	5.0	0.0	5.0	2.0
Rice bran	31.2	22.1	11.4	2	1.0
Cod liver oil	3	ę	ŝ	l ന	e e e
Tapioca flour	ŝ	ę	£	- m	3.0
Vit.Min.mix ^a	1.0	1.0	1.0	1.0	1.0
Proximate composition		Nutr	Nutrient content (%)		
Protein %	39.08	43.35	49.20	54.72	59.36
Carbohydrate %	10.50	8.88	7.239	5.248	3.77
Lipid %	17971	8.181	8.569	12.732	13.25
Gross Energy (Kcal/100g)	279.94	304.29	319.32	372.69	392.89
E/P ratio (g/Kcal)	7.162	6.709	6.489	6.810	6.618

^a Vitamin-mineral per 100g premix contained: Vitamin A 200,000 IU, Cholecalciferol 40,000 IU, Vitamin B₁₂ 80 mg Vitamin E 30 units, Vitamin K 40 mg.calcium pantochenate 100mg,nicotinamide 400mg,vitamin B₁₂ 240 mg, choline chloride 6 g, calcium 30g,manganese 1.1 g, idodine 40mg,iron mg, zinc 600mg,copper 80mg,cobalt 18 mg

Table 3: Growth performance of Channa striatus fry fed on different levels of protein for a period of 35 days

			Protein %		
	40 (D1)	45 (D2)	50 (D3)	55 (D4)	60 (D5)
Initial length (cm)	1.15±0.01ª	1.16±0.01 ª	1.14±0.01 ª	1.15±0.01 ^a	1.16±0.01 ª
Initial weight (g)	0.108±0.01 ª	0.107±0.01 ^ª	0.107±0.01 ª	0.109±0.01 ª	0.108±0.01 ª
Final length (cm)	.2.31±0.015 ^f	2.56±0.02 ^d	2.61±0.054°	3.57±0.02 ª	2.63±0.002 ^b
Final weight (g)	0.268±0.005°	0.27±0.002 ^{b,c}	0.286±0.003 ^b	0.396±0.015 ª	0.271±0.001 °
SGR (%/day)	1.13±0.011 ^d	1.163±0.002°	1.209±0.003 ^b	1.613±0.036ª	1.14±0.004 ^{с,d}
Weight gain (%)	148.77±2.370°	155.55±0.514°	165.94±1.99 ^b	267.21±10.85 ^ª	151.54±0.974°
ADG (%)	0.458±0.002°	0.479±0.002 ^{b.c}	0.509±0.006 ^b	0.824 ± 0.040^{4}	0.467±0.001 °
FCR	2.26 ± 0.040 °	2.37±0.025 ^d	2.18±0.02 ^b	1.526±0.020 ^ª	2.46±0.017°
Survival (%)	83.33±5.77ª	86.66±5.773 ª	93.33±5.773 °	96.66±5.773 ª	83.33±5.773 ª

The mean values having different superscripts in the same row are significant difference at p<0.05% level

Ē

ij

i

289

Table 4: Body composition of Channa striatus fry (dry weight basis) fed on different protein diets.

		-	Protein %	%		
	Initial	40 (D1)	45 (D2)	50 (D3)	55 (D4)	60 (D5)
Protein (%)	52.16±0.03 ^f	54.39±0.03 ^d	56.34±0.03 °	58.36±0.041 ^a	60.32±0.032ª	53.38±0.02 °
CHO (%)	1.036±0.025 ^b	1.06±0.015 ^b	1.046±0.041 ^b	0.976±0.02b ^ª	1.08 ± 0.02^{b}	0.98±0.02 ª
Lipid (%)	5.86±0.0152 ^d	6.22±0.02 ^b	7.55±0.03 ^a	6.56±0.02°	5.45±0.025 °	6.03±0.026°
Ash (%)	21.66±0.02 ^d	23.83±0.025 ^a	22.48±0.02 °	22.56±0.03 ^b	21.67 ± 0.02^{d}	20.51±0.02°
Moisture (%)	78.56±0.104ª	76.83±0.025℃	75.28±0.02 °	76.67±0.12 ^d	77.65±0.023 ^b	75.16±0.041 °
GE(Kcal/100g)	287.48±0.009 °	301.01±0.015 ^d	321.02±0.026 ^b	317.47±0.025°	321.27±0.005ª	277.46 ± 0.015^{f}
E/P	5.512±0.002°	5.53 ± 0.001^{b}	5.70±0.03 ^a	5.46±0.02 ^d	5.326±0.002 [€]	$5.2 \pm 0.01^{\text{f}}$

The mean values having different superscripts in the same row are significant difference at p<0.05% level

÷.,

, e

,

Table 5: Summary of ANOVA treatments of the effect of different levels of dietary protein on the growth performance of Channa striatus fry (the means were

compared using Duncan multiple range test).

Parameters	Source of variation	SS	df	MS	F-value	Significance
Initial length (cm)	Between groups Within groups Total	0.001 0.001 0.001	4 10 14	0.001	0.001	1.000
Initial weight (g)	Between groups Within groups Total	0.001 0.001 0.001	4 10	0.001	0.001	1.000
Final length (cm)	Between groups Within groups Total	2.808 0.002 2.810	4 10	0.702 0.001	3169.128	0.05*
Final weight (g)	Between groups Within groups Total	0.036 0.001 0.036	4 10 14	0.009 0.001	178.852	0.05*
SGR (%/day)	Between groups Within groups Total	0.499 0.003 0.502	4 10 14	0.125 0.001	408.586	0.05*
ADG (%)	Between groups Within groups Total	0.292 0.003 0.295	4 10 14	0.073	211.838	0.05*
Weight gain (%)	Between groups Within groups Total	30486.997 257.150 30744.147	4 10 14	7621.749 25.715	296.393	0.05*
FCR	Between groups Within groups Total	1.551 0.006 1.556	4 10 4	0.388 0.001	677.768	0.05*
Survival (%)	Between groups Within groups Total	360.000 1133.333 1493.333	4 10 14	90.000 113.333	0.794	0.555

Table 6: Summary of ANOVA body composition of Channa striatus fry fed on different levels of dictary protein (the means were compared using Duncan multiple range test).

Parameters	Source of variation	SS	df	MS	F-value	Significance
Protein (%)	Between groups Within groups Total	145.124 0.013 145.137	5 12 7	29.025 0.001	27789.74	0.05*
Carbohydrate (%)	Between groups Within groups Total	0.031 0.009 0.040	5 7 12	0.006	8.634	0.05*
Lipid (%)	Between groups Within groups Total	7.740 0.007 7.747	5 12 7	1.548 0.001	2786.512	0.05*
Ash (%)	Between groups Within groups Total	18.329 0.010 18.339	5 12 7	3.666 0.001	4428.458	0.05*
Moisture (%)	Between groups Within groups Total	26.340 0.060 26.400	5 12 7	5.268 0.005	1054.767	0.05*
Gross energy	Between groups Within groups Total	5265.227 0.004 5265.231	5 12 7	1053.045 0.001	3303960	0.05*
Energy/protein	Between groups Within groups Total	0.463 0.001 0.464	5 12 7	0.003	1058.118	0.05*

 \star - Statistically significant difference at (p <0.05 % level).

292

. .*****